

Elsinore Valley Municipal Water District 31315 Chaney St. Lake Elsinore, CA 92531 (951) 674-3146

# Annual Water Quality Report

Este informe contiene información muy importante sobre su agua potable. Nuestros clientes que hablan español pueden comunicarse con el Distrito al teléfono **(951) 674-3146** para recibir una traducción del informe.

## EVMWD.COM



## WE'RE HERE FOR YOU **Committed to Exceeding Expectations**

This year, our community has faced significant uncertainty as a result of the COVID-19 pandemic. One thing that our customers can be certain of is that Elsinore Valley Municipal Water District's (EVMWD) team of essential workers continue to deliver safe water to our customers 24/7. EVMWD's water is rigorously monitored and tested, allowing our customers to consume it with confidence.

We're committed to exceeding your expectations when it comes to the high-quality water we deliver to your homes and businesses. We also strive to provide exceptional customer service, timely and transparent communications and dedication to developing a secure, reliable water supply for future generations.

#### Our commitment to exceed your expectations:



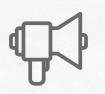
Water Quality & Delivery

Our highly trained team works 24/7 to perform more than 20,700 tests a year, ensuring that our water systems function properly and that our water meets some of the most rigorous standards in the world before it reaches your tap.



Customer Service

Our customers come first. Whether they are engaging with one of our field operations team members or our customer service representatives, customers will receive an exceptional level of service.



Communications

At EVMWD, we're proud to continuously provide timely and transparent information to the public - which has been especially critical during the pandemic and to offer plentiful opportunities for customer engagement.



Water Reliability **Projects** 

The challenges of the last year did not stop us from moving forward with several important water reliability projects that will maintain and improve our existing delivery system as we strategically meet our community's present and future water needs.

We also understand that your family may be facing financial difficulties as a result of COVID-19, and we invite you to visit our website or contact us to learn more about the numerous programs we have to help you during these uncertain times.

I invite you to read our 2020 Water Quality Report to learn more about our programs, our dedication to your water supplies, and how we carefully treat and monitor our water to ensure its safe.

We are thankful to serve our community, and we're here for you.



**Greg Thomas** 

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**General Manager Elsinore Valley Municipal Water District** 

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Elsinore Valley Municipal Water Distric

# EVMWD AT A GLANCE

Providing our customers with high-quality and reliable water service is a hallmark of our District. Maintaining underground pipelines, managing pump stations, and carefully testing our water are just a few of the many ways we ensure that water gets to your home 24/7.



## YOUR WATER QUALITY REPORT SIMPLIFIED

Water quality may seem complicated, but we're here to simplify it. Here are some tips:



Part Per MILLION or Milligrams/Liter = 1 drop in a hot tub

Maximum Contaminant Level (MCL) - The MCL is the highest level that a contaminant can safely be present in drinking water. Review the report tables. Find the contaminant that you want to evaluate. Look at what the allowable MCL is. Next, see the level listed in the data chart.

TIP: Compare the MCLs for each contaminant to the levels noted in the data charts.

What does the PHG column mean? The PHG column represents the Public Health Goals set by the California Environmental Protection Agency. They represent the level at which a contaminant has no known or expected health risks.

TIP: Compare the PHGs for each contaminant to the levels noted in the data charts. Public Health Goals can differ from MCLs and not all PHGs have a maximum level stated.

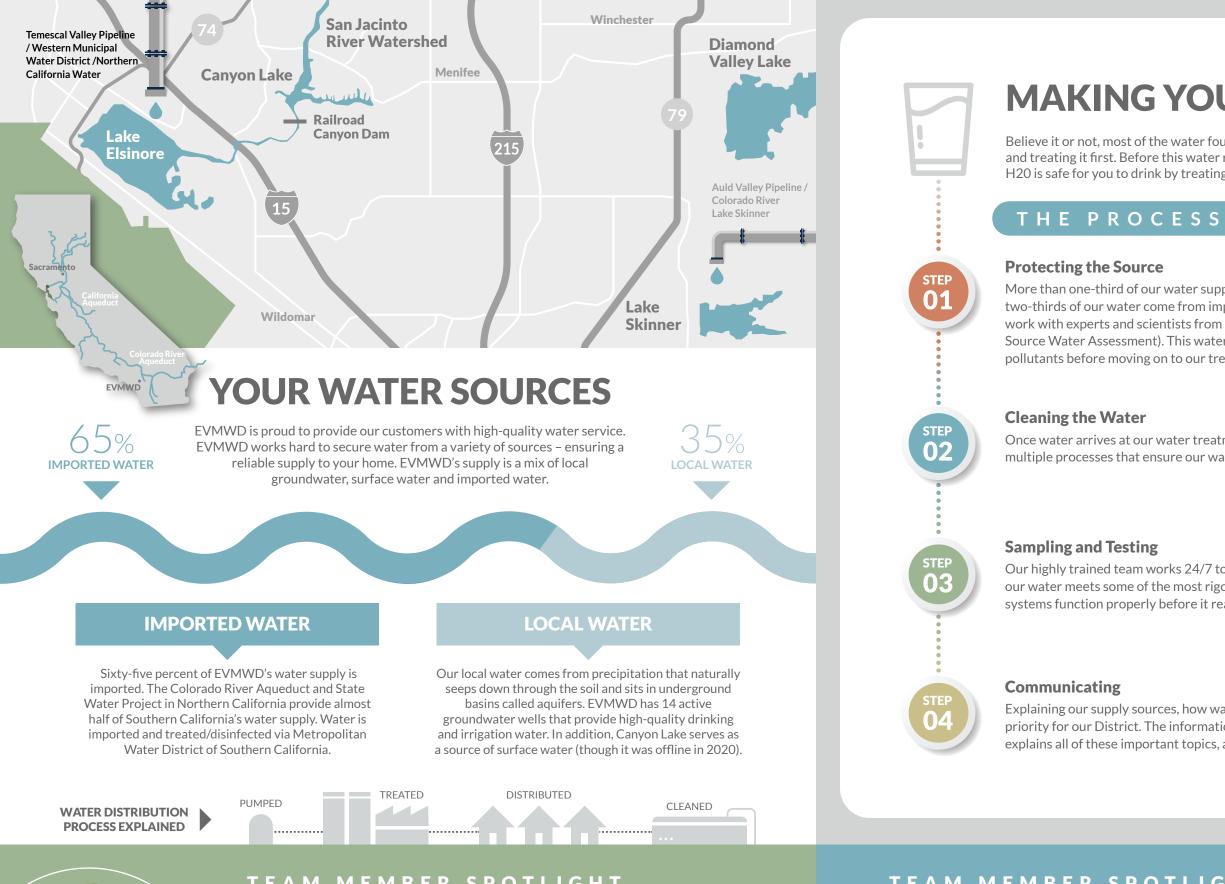




Part Per BILLION or Micrograms/Liter = 1 drop in an Olympic size swimming pool



Part Per TRILLION or Nanogram/Liter = 1 drop in a 6-acre lake





### TEAM MEMBER SPOTLIGHT

<sup>44</sup>I absolutely love my job. I've been a resident of beautiful Lake Elsinore for 26 years, Water quality is vital, and at EVMWD we ensure a trustworthy supply of water that's delivered to our customers year-round."



## TEAM MEMBER SPOTLIGHT

<sup>44</sup>Water quality is important for the health and safety of the community and provides them with access to clean drinking water. For the good of our community, and all the needs that exist from our homes to recreation to industrial/agricultural use, it is key to have commitment and dedication at the heart of what we do as water operators.

+ STEVEN GARCIA Water Production Operator



# **MAKING YOUR WATER SAFE**

Believe it or not, most of the water found in nature is generally not safe to drink without testing and treating it first. Before this water reaches your home, we ensure Mother Nature's finest H20 is safe for you to drink by treating, monitoring and testing it thousands of times per year.



More than one-third of our water supply comes from local groundwater sources; the other two-thirds of our water come from imported supplies. Before this water reaches the tap, we work with experts and scientists from the state to sample water at the source (called a Source Water Assessment). This water is tested for naturally occurring and man-made pollutants before moving on to our treatment plants.

Once water arrives at our water treatment facilities, our water quality experts rely on multiple processes that ensure our water is clean and safe.

Our highly trained team works 24/7 to perform more that 20,700 tests a year, ensuring that our water meets some of the most rigorous standards in the world and that our water systems function properly before it reaches your tap.

Explaining our supply sources, how water is monitored and how we keep it safe is a top priority for our District. The information throughout our 2020 Water Quality Report explains all of these important topics, as well as data on what we test for in your water.





## **MANAGING CONTAMINANTS IN OUR DRINKING WATER**

Providing clean, reliable drinking water to our customers is our top priority, and EVMWD is closely monitoring PFAS (per- and polyfluoroalkyl substances) in our drinking water supply. Like many communities throughout the nation, very small amounts of PFAS have been found in our water.

### Why is PFAS in drinking water?

EVMWD did not put PFAS in our water. Over time these chemicals enter our water supply through manufacturing, landfills and wastewater discharge – which are all potential sources for PFAS.

### Are PFAS harmful?

Exposure at certain levels can cause health impacts, but the exact level is still unknown. Science is evolving and the California State Water Resources Control Board (SWRCB) and U.S. Environmental Protection Agency (USEPA) continue to assess and evaluate what levels are acceptable in drinking water.

## How is EVMWD managing PFAS in our drinking water?



**Testing:** EVMWD regularly and proactively monitors the water from all of our sources to ensure it meets the state's regulations and appropriate actions, including removing water sources from

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**Treating:** Through a blend of cutting-edge strategies and proven treatment options, EVMWD is taking steps to address PFAS in our water sources. EVMWD will be installing and implementing Granulated Activated Carbon and Ion Exchange treatment processes at Canyon Lake over the next few years.

**Communicating:** We transparently communicate the latest community meetings.

Visit evmwd.com/pfas to learn more.

**PFOA, PFOS, VANADIUM:** Your drinking water contains low levels of PFOA, PFOS, and Vanadium; all falling within the current State and EPA response levels, but above notification levels established by the State. The State and EPA continue to study human health effects of these constituents, as they are known to cause adverse effects on laboratory animals, including increased liver weight, developmental and reproductive effects, immune suppression and cancer. EVMWD is providing blending of sources to reduce PFOA, PFOS, and Vanadium levels and evaluating treatment options to remove them completely to ensure state and federal mandates are met or exceeded.



## THE VALUE OF TAP WATER

Tap water is a much better value than bottled water. Spending more on bottled water doesn't guarantee better guality. Because tap water is more heavily regulated than bottled water in the United States, water agencies, like EVMWD, are required to test and monitor tap water daily.



## • TESTED DAILY

- REGULATED BY LOCAL, STATE AND FEDERAL AGENCIES
- REOUIRED TO REPORT FINDINGS
- 1 GALLON OF WATER COMES OUT TO \$0.006 (LESS THAN A PENNY!)

## **Compared to tap water, bottled water is:**

\*According to the California Department of Public Health Water Quality Analyses Database



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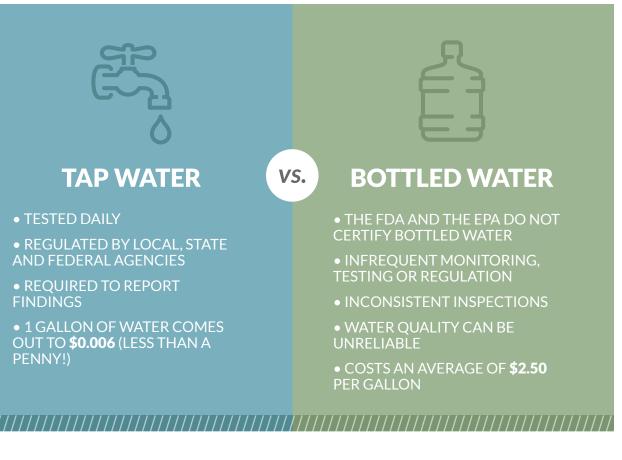


Investing in our pipelines, facilities and people ensures service reliability and fiscal stability for the future. Key projects coming soon to EVMWD include:

#### **Canyon Lake Water Treatment** Plant, Phase I Improvements

Installing granulated activated carbon and ion exchange treatment processes at Canyon Lake to treat for PFAS/PFOS.

Replacing existing aged Mayhew Well to improve the water supply reliability within the Temescal Service Area.







3.2x MORE LIKELY TO VIOLATE **HEALTH STANDARDS** 



#### **Temescal Wells Replacement Project**



Construction of an interconnection between EVMWD and Eastern Municipal Water District at Goetz Rd. to provide an additional water source to the Canyon Lake and Canyon Hills Area.



## **About Your Water Quality Report**

Enclosed for your review is our accumulation of 2020 water quality testing. Testing frequency and water quality levels are set by the State Water Quality Control Board, Division of Drinking Water (SWRCB-DDW). The Elsinore Valley Municipal Water District's goals are to provide safe drinking water to its customers and follow the policies and procedures of the State of California and U.S. Environmental Protection Agency (U.S. EPA). EVMWD maintains chlorine disinfectant residuals in the drinking water as mandated by the (SWRCB-DDW) and U.S. EPA.

Assessments of drinking water sources were completed as required by the State Water Control Board, Division of Drinking Water. The sources are most vulnerable to the following activities not associated with any detected contaminants: airports, gravel mining operations, machine shops, maintenance yards, septic systems, sewer collection systems, and transportation corridors. A copy of the complete assessment is available at EVMWD.

#### Important Facts from the U.S. EPA About Drinking Water

Sources of drinking water (both tap water and bottled water) include rivers, lakes, streams, ponds, reservoirs, springs, and wells. As water travels over the surface of the land or through the ground, it dissolves naturally occurring minerals and, in some cases, radioactive material, and can pick up substances resulting from the presence of animals or from human activity.

#### Contaminants that may be present in untreated sources may include:

**Primary Contaminants** adversely affect public health. **Secondary Contaminants** may cause cosmetic effects (such as skin or tooth discoloration) or aesthetic effects (such as taste, odor, or color) in drinking water.



 (Primary), such as viruses
 and bacteria, that may come from sewage treatment
 plants, septic systems,
 agricultural livestock
 operations, and wildlife.

**Microbial contaminants** 

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Pesticides and herbicides
 (Primary), which may come from a variety of sources such as agriculture, urban storm water runoff, and residential uses.



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**Secondary)**, such as salts and metals, that can be naturally occurring or result from urban storm water runoff, industrial or domestic wastewater discharges, oil and gas production, mining, or farming.

Inorganic contaminants (Primary &

Organic chemical contaminants (Primary), including synthetic and volatile organic chemicals, that are byproducts of industrial processes and petroleum production, and can also come from gas stations, urban storm water runoff, agricultural application, and septic systems.

## Radioactive contaminants (Primary),

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**Request a Summary** 

of the Assessment

Contact Mike Ali, Water

Quality Administrator, at

(951) 674-3146 x8256 or hali@evmwd.net

> which can be naturally occurring or the result of oil and gas production, and mining activities.

In order to ensure water is safe to drink, the United States Environmental Protection Agency (EPA) and the State Water Resources Control Board (SWRCB) prescribe regulations that limit the amount of certain contaminants in water provided by public water systems. SWRCB regulations also establish limits for contaminants in bottled water to provide the same protection for public health.

## Water Quality Terms

**AVERAGE:** The average reported in the data is the combined result of multiple collection samples.

MAXIMUM CONTAMINANT LEVEL (MCL): The highest level of a contaminant that is allowed in drinking water. Primary MCLs are set as close to the Public Health Goals (PHG) (or MCLGs) as is economically and technologically feasible. Secondary MCLs are set to protect the odor, taste, and appearance of drinking water.

**MAXIMUM CONTAMINANT LEVEL GOAL (MCLG):** The level of a contaminant in drinking water below which there is no known or expected risk to health. MCLGs are set by the U.S. Environmental Protection Agency (EPA).

MAXIMUM RESIDUAL DISINFECTANT LEVEL (MRDL): The highest level of a disinfectant allowed in drinking water. There is convincing evidence that the addition of a disinfectant is necessary for control of microbial contaminants. MAXIMUM RESIDUAL DISINFECTANT LEVEL GOAL (MRDLG): The level of a drinking water disinfectant below which there is no known or expected risk to health. MRDLGs do not reflect the benefits of the use of disinfectants to control microbial contaminants.

NOTIFICATION LEVEL (NL): A health-based advisory level established by the state for chemicals in drinking water that lack maximum contaminant levels (MCLs).

**PRIMARY DRINKING WATER STANDARD (PDWS):** MCLs and MRDLs for contaminants that affect health along with their monitoring and reporting requirement and water treatment requirements.

**PUBLIC HEALTH GOAL (PHG):** The level of a contaminant in drinking water below which there is no known or expected risk to health. PHGs are set by the California Environmental Protection Agency.

**REGULATORY ACTION LEVEL (AL):** The concentration of a contaminant which, if exceeded, triggers treatment or other requirements that a water system must follow.

**TREATMENT TECHNIQUE (TT):** A required process intended to reduce the level of a contaminant in drinking water.

**TURBIDITY:** Is a measure of the cloudiness of the water, and it is a good indicator of the effectiveness of our filtration system.

**UNREGULATED CONTAMINANT MONITORING RULE (UCMR):** Helps USEPA and the State Water Resources Control Board to determine where certain contaminants occur and whether the contaminants need to be regulated.

The State allows us to monitor for some contaminants less than once per year because the concentrations of these contaminants do not change frequently. Some of our data, though representative is more than one year old.

## Important Info from the EPA on Drinking Water

Drinking water, including bottled water, may reasonably be expected to contain at least small amounts of some contaminants. The presence of contaminants does not necessarily indicate the water poses a health risk. More information about contaminants and potential health effects can be obtained by calling the U.S. EPA's Safe Drinking Water Hotline at 1-800-426-4791 or visiting the EPA's web site at www.epa.gov. Trace chemicals are measured in parts per million (ppm), which is the same as milligrams per liter (mg/L). Some constituents are measured in parts per billion (ppb).

Some people may be more vulnerable to contaminants in drinking water than the general population. Those who may be particularly at risk include cancer patients, organ transplant recipients, people with HIV-AIDS or other immune system disorders, as well as some elderly individuals and infants. These people should seek advice about drinking water from their health care providers. U.S. Centers for Disease Control & Prevention (CDC) guidelines on appropriate means to lessen the risk of infection by Cryptosporidium and other microbial contaminants are available from the Safe Drinking Water Hotline at 1-800-426-4791 or visit water.epa.gov/drink/hotline.

#### ► ARSENIC

Your drinking water contains low levels of arsenic that fall within state and federal health-based standards and are below thresholds that would require corrective action. To protect public health, the U.S. Environmental Protection Agency sets maximum levels for contaminants based on the best available treatment technology to remove them from drinking water. The EPA continues to research the health effects of low levels of arsenic, a mineral known to cause cancer in humans at high concentrations that is linked to other health effects such as skin damage and circulatory problems. In 2008, EVMWD completed construction on the \$8 million Back Basin Groundwater Treatment facility that removes arsenic and other naturally occurring contaminants that are often found in groundwater.

#### ► LEAD

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Since 2017, public schools have had the option of requesting local water agencies collect water samples to test for lead. New regulations required local water agencies to test lead levels by July 1, 2019, at all K-12 schools constructed before 2010. During 2018-19, EVMWD completed drinking water lead testing at all K-12 public schools in the service area. None of the schools exceeded the Action Level for Lead in tap waters. If present, elevated levels of lead can cause serious health problems, especially for pregnant women and young children. Lead in drinking water is primarily from materials and components associated with service lines and home plumbing. EVMWD is responsible for providing high-quality drinking water, but cannot control the variety of materials used in plumbing components. When your water has been sitting for several hours, you can minimize the potential for lead exposure by flushing your tap for 30 seconds to 2 minutes before using water for drinking or cooking. If you are concerned about lead in your water, you may wish to have your water tested.

Information on lead in drinking water, testing methods, and steps you can take to minimize exposure is available from the Safe Drinking Water Hotline, toll free at 1-800-426-4791 or at www.epa.gov/safewater/lead.

#### ► SALT

One of the most important issues facing water supplies throughout Southern California today is salinity. Total Dissolved Solids (TDS), also known as salinity, is the concentration of dissolved mineral salts such as calcium, magnesium, sodium sulfate, and chloride. Local water supplies and recycled water have continued to show an increase in salt content. Though these salts are viewed as an aesthetic standard by the State Water Resources Control Board, too much salt can negatively impact our local water sources, agriculture, and our environment. EVMWD is currently exploring options on how to meet state-mandated requirements to eliminate the overabundance of these salts.

#### **RADON**

Radon is a naturally occurring gas formed from the normal radioactive decay of uranium. Radon has been detected in our finished water supply. There are no regulatory limits prescribed for radon levels in drinking water – the pathway to radon exposure occurs primarily through its presence in the air. Exposure over a long period of time to air containing radon may cause adverse health effects. If you are concerned about radon in your home, testing is inexpensive and easy. For more information, call your state radon program (1-800-745-

7236), the National Safe Council's Radon Hotline (1-800-SOS-RADON), or the EPA Safe Drinking Water Act Hotline (1-800-426-4791).

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## **Consumer Confidence Report 2020**

In 2020, Elsinore Valley Municipal Water District collected approximately 2,485 drinking water samples at various locations throughout the water system, yielding 20,700 test results that are summarized below. As per SWRCB-DDW guidelines, the tables include only those contaminants that were detected during 2020 or prior sampling years as applicable. It is important to note that the presence of these contaminants, as detected in the water, does not necessarily indicate that the water poses a health risk. We are pleased to report that no drinking water violations occurred during the 2020 compliance period.

		-	-		-		-		-		m Water	-		ary	
											RM BACTERIA				
Microbiological Co	No. of months in violation	MCL							PHG, MCLG	Typical Source of Bacteria					
Total Coliform Bac (State Total Colifor			1.2%	5	0	More than 5% of monthly samples are positive							0	Naturally present in the environment	
E. coli (State Total C		0		sample and coliform			detect to	otal coliform and ei	ther sample also	Human and animal fecal was					
E. coli (Federal Rev Coliform Rule)	ised Total		0%		0	or system	repeat	t samples f	ollowing	positive and either g E. coli-positive ro epeat sample for E	utine sample or	Human and animal fecal wast			
DI	STRIBUTI	ON SI	YSTEM R	ESULTS FO	OR DISINFE	CTANT	RESIDUA	ALS A	AND DIS	SINFE	CTION BYPR	DUCTS		CHLORATE NOTIFICATION:	
Chemical or C (and reporti			Sample Year	Highest LRAA*	Range of Detections				MCL,MRDL Violation		Typical Sou	ırce of Contami	nant	Chlorate concentrations in some UCMR samples were found above State Notification Level of 800	
Total Trihalomethan	es-TTHMs (p	pb) 2	2020	38.2	20.1-52	80	NA		No	E	Byproduct of drii	nking water chlo	rination	ppb. Use of Environmental Sources of Chlorate include agricultural	
Haloacetic Acids-H	IAA5 (ppb)	2	2020	10.4	4.6-14	60	NA		No	E	Byproduct of dri	nking water chlo	rination	defoliant or desiccant, disinfection byproduct, and use in production o	
Free Chlorine (ppm) 2020				RAA=1.6	0.01-4.2	MRDL=			No			-		chlorine dioxide. Health Effects of	
Fotal Chlorine (ppn							inking water disinfectant added for treatment inking water disinfectant added for treatment B-11-001 (Jan-2012)								
iotal eniornie (ppi		2											treatment		
Lead and Copper Rule (and reporting units)	d Sample samples centile level exceeding AL PHG DLR Schools Lead						Lead Testing		pical Sour	rce of Contaminant					
Lead (ppb)	2019	76	ND		0	15	0.2	5		-2019 (2 equests	27), 2020 ;)			ld water plumbing systems; discharg ;; erosion of natural deposits	
Copper (ppm)	2019	76	0.25	5	0	0.3	0.05	5 N/A					old plumbing systems; erosion of om wood preservatives		
				DI	STRIBUTIO	N SYSTI	EM RESU	JLTS	FOR OT	HER I	PARAMETERS	;			
Chemical or Constituent (and reporting units) Sample Year Average Level Detected Detections								MCL or (SMCL)	РНС	3	Typical Source of Contaminant				
Heterotrophic Plate Count (HPC) 2020					13.5	0-555		TT NA		Naturally pro	esent in the enviro				
furbidity (Distributi	on System), N	NTU (a.1	L)	2020	0.2	0.07-2.25			(5) NA		Soil Runoff				
Color				2020	1.4		0-25		(15)	NA	Naturally oc	Naturally occuring organic materials			
рН	2020	7.9		6.81-8.96		NA	NA								
Temperature				2020	22.4		11.9-41.3		NA	NA					
Alkalinity total (ppm) 20					0		83-150		NA NA	NA NA					
Odor (Tons)				2020	0		0-3		NA	NA					
Federal UCMR 4 Analyte detections									Units				stem		
												Range		Average	
Bromide									ug/L		_	30-310	211		
Bromochloroacetic acid									ug/L			ND-4.3	2.1		
Bromodichloroace									ug/L ug/L		-	ND-2.4 ND-2.1	<u> </u>		
Chlorodibromoace Dibromoacetic acio									-			ND-2.1 ND-6.2	0.7		
Dibromoacetic ació Dichloroacetic ació									ug,						
									ug,			ND-5.7 2.5			

AlkA LINITY (ICIAL) SC ACO3     ppm     NA     NA<	PRIMARY DRINKING WATER STANDARDS											
Construct (Unity)     Orac     Image: Image							SURFACE	E WATER &	GROUNI	NDWATER		
Number     Number </th <th>Constituent (Units)</th> <th>Units</th> <th>MCL</th> <th>PHG</th> <th>DLR</th> <th></th> <th>MWD-Mills TVP &amp; Flagler</th> <th>MWD- Skinner/ Auld Valley</th> <th colspan="2">Elsinore Temescal Ground- Ground-</th> <th>IVICE</th> <th>Major Sources in Drinking Water</th>	Constituent (Units)	Units	MCL	PHG	DLR		MWD-Mills TVP & Flagler	MWD- Skinner/ Auld Valley	Elsinore Temescal Ground- Ground-		IVICE	Major Sources in Drinking Water
Nome     Nome     Nome     Nome     Nome     Nome     Nome     Nome       ARENNE     gab     1     0 </td <td>Turbidity (Treatment Plant Effluent)</td> <td>NTU</td> <td>TT=1</td> <td>NA</td> <td>NA</td> <td></td> <td>0.09</td> <td>RITY 0.09</td> <td></td> <td></td> <td>No</td> <td>Soil runoff</td>	Turbidity (Treatment Plant Effluent)	NTU	TT=1	NA	NA		0.09	RITY 0.09			No	Soil runoff
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main and participand partiteparticipand participand participand participand participa			+ +			Average	0.6	0.7	0.3	0.3		strong teeth; discharge from fertilizer and aluminum factories
PLACEMONITY         PIPE         PIPE         Parking         Project         PLACEMONITY         PLACEMONITY <td></td> <td></td> <td></td> <td></td> <td></td> <td>Average</td> <td>2.0</td> <td>ND</td> <td>2.0</td> <td>2.2</td> <td></td> <td>deposits erosion</td>						Average	2.0	ND	2.0	2.2		deposits erosion
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CODE SETA         Part Part Part Part Part Part Part Part		pCi/L	15	MCLG=0	3	Range	ND-10.1	ND - 3	ND-11.3			Erocion of natural denosits
OMODE INDeckParty<					-	Range	ND-7.41	ND - 5	ND	12.8		
OuthorneyDeckPoint						Average Range	ND ND-3.55	ND - 2	ND-5.35		5	
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Signed Fuldwarets Adds PMAS     Option     Option     Option     Option     No.     Option     Option <td>Total Trihalomethanes (TTHM)</td> <td></td> <td></td> <td></td> <td></td> <td>Range Average</td> <td>14-22 18</td> <td>13-24 23</td> <td>NA NA</td> <td>NA NA</td> <td></td> <td></td>	Total Trihalomethanes (TTHM)					Range Average	14-22 18	13-24 23	NA NA	NA NA		
Brends     Image	Sum of Five Haloacetic Acids (HAA5)	ppb	60	NA	1.0	Range Average	2.2 - 14 9.1	3.5 - 12 8.5	NA NA	NA NA		Byproduct of drinking water chlorination
TOTAL ORGANG CABBO MIDOR)     open     T     NA     Object     Provide and any and any and any	Bromate	ppb	10	0.1	1.0	Range	ND - 12 4.3	ND - 5.6 2.5	NA NA	NA NA		
SECONDA RY STANDARDS A ESTHETICS TANDARDS         ALUMINUM         SECONDA RY STANDARDS         ALUMINUM         Second Colspan="2">Second Colspan="2">Second Colspan="2">Second Colspan="2">Second Colspan="2">Second Colspan="2"           ALUMINUM         ppi         Second Colspan="2"         Second Colspan="2" <th< td=""><td>TOTAL ORGANIC CARBON (TOC)</td><td></td><td>ТТ</td><td>NA</td><td>0.30</td><td>Range</td><td>ND-3.1</td><td>1.9 - 2.6</td><td>ND-0.6</td><td>ND-4</td><td></td><td></td></th<>	TOTAL ORGANIC CARBON (TOC)		ТТ	NA	0.30	Range	ND-3.1	1.9 - 2.6	ND-0.6	ND-4		
ALMMOVEIponEndAndNoParageRolNo			SECO	NDA	<b>RY</b> S	STAND	DARDS	- AEST	THETI	IC STAI		RDS
CHORE     Field     Field    Field    <	ALUMINUM	ppb	200	600	50	Average	ND	108	ND	ND	No	
COLORUnitsUnitsNA	CHLORIDE	ppm	500	NA	NA	Range Average	60-70 65.5	96 96.0	54-150 91.0	45-47 46.0		
Family Sentence (Mark) Book Sentence (Mark) ROMoppipp<		Units	15	NA	NA	Range Average	0-3	1-2 2	0	0-0		Naturally-occurring organic materials
and and probability proba		ppb	500	NA	NA	Range Average	0	ND ND	0-90 11.3	0-0		Municipal and industrial waste discharges
MAGAMESE         pp/l         f<         f         f         <			300	NA	100	Range	ND ND	ND ND	ND-610 ND	ND ND		
ODOR THRESHOLD         Units         3         NA         1         Page MO2         Page MO2         Page MO2         Page MO2         Note MO2 <td></td> <td></td> <td></td> <td></td> <td></td> <td>Range</td> <td>ND</td> <td>ND</td> <td>ND-23</td> <td>ND-31</td> <td></td> <td></td>						Range	ND	ND	ND-23	ND-31		
Selectric Conduction Conduction System         Open         Size         Available and available available and available and available and available					-	Range	ND-2	2	ND	ND		
SULFATE         pm         pm <t< td=""><td></td><td></td><td></td><td></td><td></td><td>Range</td><td>439-1400</td><td>796 - 956</td><td>6 480-1100</td><td>0 640-760</td><td>)</td><td></td></t<>						Range	439-1400	796 - 956	6 480-1100	0 640-760	)	
John         John <th< td=""><td></td><td></td><td></td><td></td><td></td><td>Range</td><td>58-75</td><td>152 - 208</td><td>3 78-240</td><td>110-110</td><td>)</td><td></td></th<>						Range	58-75	152 - 208	3 78-240	110-110	)	
IDIAE DISCURSION METABOR (199)     IDIA     IDIAE     Average     Add     Average     Add     Average     Add     Add <t< td=""><td></td><td></td><td></td><td></td><td></td><td>Range</td><td>318-400</td><td>308-608</td><td>282-614</td><td>340-472</td><td>2</td><td>• • • •</td></t<>						Range	318-400	308-608	282-614	340-472	2	• • • •
No. Bit Mathematical system)         N/I         N/I         N/I         Average         0.0         0.0         0.0         0.00000000000000000000000000000000000						Range	0.18-0.56	ND-0.38	0-0.5	0-2.09		
CHALLINTY (TOTAL) AS CACO3         ppm         NA         NA <t< td=""><td></td><td></td><td></td><td></td><td>Ì</td><td></td><td></td><td></td><td></td><td>0.3</td><td>No</td><td></td></t<>					Ì					0.3	No	
ALKALINITY (TOTAL) AS CACO3         ppm         NA         <						G	GENERAL M	MINERALS	_S			
CACIUM     pp     pp     NA	ALKALINITY (TOTAL) AS CACO3	ppm	NA	NA	(1)	Range Average	74-293 178	78-126 113	76-164 131	148		Runoff/leaching of natural deposits; carbonate, bicarbonate, hydroxide, and occasionally borate, silicate, and phosphate
HARDNESS [TOTAL] AS CACO3     ppm     NA     NA     NA     NA     NA     Parage     64-22     21/2     32/34     49-37     Company and all magnetining depolits; using on polyalent calons, gen all magnetining and all magnet and all magnetinin	CALCIUM	ppm	NA	NA	(0.1)	Range Average	32-45 39	52 - 72 68	28-85 68	62-68 65	NA	Runoff/leaching from natural deposits
MAGNESUM         ppm         NA	HARDNESS (TOTAL) AS CACO3	ppm	NA	NA	(1)	Range Average	64-622 303	211-273 242	3 33-284 224	198-317 231	7	Runoff/leaching from natural deposits; sum of polyvalent cations, gener- ally magnesium and calcium present in the water
POTASSIUM         ppm         NA						Range	9.7-14 11.9	20-26 24	5.5-17 13	13-15 14		
SODUM         ppm         NA         NA <th< td=""><td></td><td></td><td></td><td></td><td></td><td>Average Range</td><td>2.5-3.4</td><td>4.0 - 4.8</td><td>0-3.2</td><td>1.7-2</td><td></td><td></td></th<>						Average Range	2.5-3.4	4.0 - 4.8	0-3.2	1.7-2		
Universe         Data Part of the construction of the construling the construction of the construction of the cons						Range	51-56	76 - 98	50-120	37-44		
BOKON         ppb         NL=100         NA         100         Average         130         ND         ND         NA         Kundividacing from natural deposits industrial wastes           CHLORATE         ppb         NL=800         NA         20         Rame         27         34         110-100         NA         NA           CHROMIUM VI         ppb         NA         0.02         1         Rame         ND         ND         ND         NA         NA           VANADIUM         ppb         NL=50         NA         30         Rame         ND         ND         ND         ND         NA           VANADIUM         ppb         NL=50         NA         33         Zamage         2.5         4.2         NA         NA         Naturally-occurring: industrial wasted discharge           VANADIUM         ppt         NL=50         NA         4         Range         2.5         4.2         NA         NA         Naturally-occurring: industrial wasted discharge           VENDROOCTANOIC ACID (PFOA)         ppt         NL=50         NA         4         Range         ND<30         ND         ND         3.2         Naturally-occurring: industrial wasted discharge           PERFLUOROCALACID (PFOA) <t< td=""><td></td><td></td><td></td><td></td><td></td><td>UNREG</td><td>GULATED C</td><td>CONTAMIN</td><td>INANTS</td><td></td><td>INPA</td><td></td></t<>						UNREG	GULATED C	CONTAMIN	INANTS		INPA	
CHLORATE         ppb         NL<800         NA         20         Rarge         27         34         10-100         NA         NA         NA           CHROMIUM VI         ppb         NA         0.02         1         Rarge         ND	BORON	ppb	NL=1000	NA	100	Average	188	130	ND	ND	NA	Runoff/leaching from natural deposits; industrial wastes
CHROMIUM VI         ppb         NA         0.02         1         Range Average         ND         ND <thn< td=""><td>CHLORATE</td><td>ppb</td><td>NL=800</td><td>NA</td><td>20</td><td>Range Average</td><td>27 27</td><td>34 34</td><td>110-1600 422</td><td>0 NA NA</td><td></td><td>Byproduct of drinking water chlorination; industrial processes</td></thn<>	CHLORATE	ppb	NL=800	NA	20	Range Average	27 27	34 34	110-1600 422	0 NA NA		Byproduct of drinking water chlorination; industrial processes
VANADUM         ppb         NL         Solution         NL         Solution         ND         ND <td>CHROMIUM VI</td> <td>ppb</td> <td>NA</td> <td>0.02</td> <td>1</td> <td>Range Average</td> <td>ND ND</td> <td>ND ND</td> <td>ND-1.2 ND</td> <td>ND ND</td> <td></td> <td>Runoff/leaching from natural deposits; discharge from industrial wastes</td>	CHROMIUM VI	ppb	NA	0.02	1	Range Average	ND ND	ND ND	ND-1.2 ND	ND ND		Runoff/leaching from natural deposits; discharge from industrial wastes
N-Nitrosodimethylamine (NDMA)         ppt         NL=10         3         (2)         Range Average         2.5         4.2         NA         NA         Mage         Byproduct of drinking water chloramination; industrial processes           PERFLUOROOCTANOIC ACID (PFOA)         ppt         NL=5.1         NA         4         Range         2.3-6.9         ND         ND-6.9         4.1-9.1         Mage         Aurage         2.3-6.9         ND         ND         7.7         NA	VANADIUM	ppb	NL=50	NA	3	Range Average	ND ND	ND ND	8.9-71 27	ND ND		Naturally-occurring; industrial waste discharge
PERFLUOROLICYL AND POLYFLUOROALKYL SUBSTANCES (PFAS)         PERFLUOROOCTANOIC ACID (PFOA)       ppt       NL =5.1       NA       4       Average       2.3 é.9       ND       ND -5.9       4.1 state       Industrial chemical factory discharges; runoff/leaching from landfills; used in fire-retarding fooms and various industrial processes         PERFLUOROCTANE SULFONIC ACID (PFOS)       ppt       NL=6.5       NA       4       Range       ND -3.6       ND -4.2       2.3 ks       issed in fire-retarding fooms and various industrial processes         PERFLUOROBUTANESULFONIC ACID (PFBS)       ppt       NL=500       NA       4       Range       ND -5.2       ND       ND -2.4       2.3 ks       ND       ND       A       Average       3.0       ND       ND -3.4       ND       ND       ND       A       Average       ND       ND       ND -3.4       ND	N-Nitrosodimethylamine (NDMA)	ppt				Range Average	2.5 2.5	4.2 4.2	NA NA	NA NA	NA	Byproduct of drinking water chloramination; industrial processes
PERFLUOROOC TANOIC ACID (PFOA)         ppt         NL = 5.1         NA         4         Average         4.6         ND         ND         7.7         NA           PERFLUOROOC TANOIC ACID (PFOS)         ppt         NL=6.5         NA         4         Range         ND         ND         ND         3.2         NA           PERFLUOROOC TANOIC ACID (PFOS)         ppt         NL=6.5         NA         4         Range         ND-3.6         ND         ND         3.2         NA           PERFLUOROBUTANESULFONIC ACID (PFBS)         ppt         NL=500         NA         4         Range         ND         ND-2.6         2.3.3.5         NA           PERFLUOROHEZANE SULFONIC ACID (PFHA)         ppt         NA         NA         4         Range         ND         ND         ND-2.2         NA           PERFLUOROHEXANOIC ACID (PFHA)         ppt         NA         NA         4         Range         ND         ND         ND-17         2.8.4.5         ND           (CCPP) IGL ACID (PFHA)         ppt         NA         NA         4         Range         0.7.4.2         2.4         ND-3.2         NA           (CCPP) IGL ACID (PFHA)         ppt         NA         NA         Range         0.85 - 2						<b>LKYL AND</b>	<b>D POLYFLU</b>	UOROALK	KYL SUBST	STANCES (I	(PFAS)	
PERFLUOROCC TARE SULPONIC ACID (PF0s)         ppt         NL = 50         NA         4         Average         2.6         ND         ND         3.2         NA           PERFLUOROBUTANESULFONIC ACID (PFBs)         ppt         NL = 500         NA         4         Range         ND - 5.2         ND         ND - 2.5         NA           PERFLUOROHEPTANOIC ACID (PFHpA)         ppt         NA         NA         4         Range         ND         ND         ND         ND - 3.4         ND - 3.4           PERFLUOROHEPTANOIC ACID (PFHpA)         ppt         NA         NA         4         Range         ND ND         ND ND - 1.0         ND - 3.4         ND - 3.4           PERFLUOROHEXANE SULFONIC ACID (PFHpA)         ppt         NA         NA         4         Range         ND - 3.2         ND         ND - 3.4         ND - 3.4           PERFLUOROHEXANOIC ACID (PFHxA)         ppt         NA         NA         4         Range         2.1         ND - 5.2         NA           (CCPP) (as CaCO3)         ppt         NA         NA         A         Range         0.52         2.078 - 11         NA         NA         A           (CCPP) (as CaCO3)         A.I.         NA         NA         NA         A			+ +			Average	4.6	ND	ND	7.7		
$ \begin{array}{c c c c c c c c c c c c c c c c c c c $						Average	2.6	ND	ND	3.2		4
PERFLUOROHEPTANOIC ACID (PFHpA)         ppt         NA         NA         VA         VA <thva< th="">         VA         VA</thva<>						Average	3.3	ND	ND	2.5	NA	4
PERFLUCIONEAXINE SOLFONIC ACID         ppt         NA         NA         A         Average         2.1         ND         2.4         3.5         NA           PERFLUOROHEXANOIC ACID (PFHxA)         ppt         NA         NA         4         Range         3.7-6.2         2.4         ND         3.5         NA           PERFLUOROHEXANOIC ACID (PFHxA)         ppt         NA         NA         4         Range         3.7-6.2         2.4         ND         5.2         NA           Colum Carbonate Precipitation Potential (CCPP) (as CaCO3)         ppm         NA         NA         Range         0.85-2.2         0.78-11         NA         NA         NA           Corrosivity (as Aggressiveness Index)         A.I.         NA         NA         Range         11.4-12.2         11.3-12.1         11.1-17         Elemental balance in water; affected by temperature, other factors           Corrosivity (as Saturation Index)         SI         NA         NA         Range         0.27-0.28         0.39-0.73         NA         NA         Elemental balance in water; affected by temperature, other factors           pH         pH units         NA         NA         Range         0.27-0.28         0.39-0.73         NA         NA         Elemental balance in water; affected by temperatu		ppt	NA	NA	4	Average	ND	ND	ND	2.2		4
PERFLUOROHEXANOIC ACID (PFHxA)       ppt       NA       A       Range       3.7-6.2       2.4       ND-3.2       3.4-6.3         PERFLUOROHEXANOIC ACID (PFHxA)       ppt       NA       A       Range       3.7-6.2       2.4       ND<3.2		ppt	NA	NA	4	Average	2.1	ND	2.4	3.5		4
Calcium Carbonate Precipitation Potential (CCPP) (as CaCO3)       ppm       NA       NA       Range       0.85 - 2.2       0.78 - 11       NA       NA       NA       NA         Corrosivity (as Aggressiveness Index)       A.I.       NA       NA       NA       NA       Range       1.6       6.4       NA       NA <td></td> <td>ppt</td> <td>NA</td> <td>NA</td> <td>4</td> <td>Range</td> <td>3.7-6.2</td> <td>2.4 2.4</td> <td>ND-3.2</td> <td>3.4-6.3</td> <td></td> <td></td>		ppt	NA	NA	4	Range	3.7-6.2	2.4 2.4	ND-3.2	3.4-6.3		
Corrosivity (as Aggressiveness Index)         A.I.         NA         NA         NA         NA         NA         NA         NA           Corrosivity (as Aggressiveness Index)         A.I.         NA         NA         NA         NA         NA         NA         NA           Corrosivity (as Saturation Index)         SI         NA         NA         NA         NA         NA         NA         NA         NA           pH         pH units         NA         NA         NA         NA         NA         NA         NA         Elemental balance in water; affected by temperature, other factors           pA         pH units         NA         NA         NA         NA         NA         NA         NA           pCi/L         NA         NA         NA         Range         6.27 - 0.28         0.39 - 0.73         NA         NA         NA         Elemental balance in water; affected by temperature, other factors         Elemental balance in water; affected by temperature, other factors           pH         pH units         NA		ppm	NA	NA	NA		0.85 - 2.2	0.78 - 11			NA	Elemental balance in water; affected by temperature, other factors
Corrosivity (as Saturation Index)         SI         NA         NA         NA         NA         NA         II.8         II.3         II.4         NA         NA           pH         pH units         NA         N			NA			Range	11.4-12.2	11.3-12.1	l 11.1-11.9	9 11.1-11.7	7	
pH         pH units         NA         <						Range	0.27 - 0.28	0.39 - 0.73	3 NA	NA		
PH     PH     INA     INA     INA     INA     INA     Average     7.6     7.8     7.7     7.2     INA     Not applicable       RADON     pCi/L     NA     NA     NA     Average     7.6     7.8     7.7     7.2     NA     Not applicable       RADON     pCi/L     NA     NA     NA     Average     450     ND     179-1710     1660-2370     6as produced by the decay of naturally-occurring uranium in soil and wa						Range	6.97-8.5	7.36-8.15	5 6.84-8.91	1 6.77-7.58	8	
RADON PLI/L NA NA 100 Average 450 ND 712 2015 NA Gas produced by the decay of naturally-occurring uranium in soil and wa	-	-				Average Range	7.6 ND-954	7.8 ND	7.7 179-1710	7.2 0 1660-2370	NA 70	Not applicable           Gas produced by the decay of naturally-occurring uranium in soil and water
	-											

Al: Aggressiveness Index AL: Action Level CaCO3: Calcium Carbonate CFU: Colony-Forming Units DBP: Disinfection Byproducts DDW: Division of Drinking Water DLR: Detection Limits for Purposes of Reporting GPG: Hardness conversion as grains per gallon - 1 GPG = 17.1 ppm as CaCO3 LRAA: Locational Running Annual Average: highest LRAA is the highest of all Locational Running Annual Averages calculated as average of all samples collected within a MFL: Million Fibers per Liter MRL: Method Reporting Level **µS/cm:** microSiemen per centimeter; or micromho per centimeter (µmho/cm)

Haloacetic acids 5 / HAA5

Haloacetic acids 6 / HAA6

Haloacetic acids 9 / HAA9

Total Organic Carbon / TOC

Monobromoacetic acid

Manganese (total)

Tribromoacetic acid

Trichloroacetic acid

MBAS: Methylene Blue Active Substances MCL: Maximum Contaminant Level MCLG: Maximum Contaminant Level Goal MRDL: Maximum Residual Disinfectant Level MRDLG: Maximum Residual Disinfectant Level Goal NA: Not Analyzed/Not Applicable ND: Not Detected above State DLR

NTU: Nephelometric Turbidity Units pCi/L: picoCuries per Liter PHG: Public Health Goal **ppb:** parts per billion or micrograms per liter (µg/L) ppm: parts per million or milligrams per liter (mg/L)
ppq: parts per quadrillion or picograms per liter (pg/L)
ppt: parts per trillion or nanograms per liter (ng/L) RAA: Running Annual Average; highest RAA is the highest of all Running Annual Averages calculated as average of all the samples collected within a 12-month period Range: Results based on minimum and maximum

ND-12

ND-19

ND-24

ND-83

ND-1

0.43-7.1

ND-4

ND-2.7

ug/L

ug/L

ug/L

ug/L

ug/L

mg/L

ug/L

ug/L

NL: Notification Level to SWRCB

RTCR: Revised Total Coliform Rule  $\textbf{SCML:} Secondary \ Contaminant \ Level \ (Aesthetic$ Standard SI: Saturation Index (Langelier) SWRCB: State Water Resources Control Board TON: Threshold Odor Number TT: Treatment Technique is a required process intended to reduce the level of a contaminant in drinking water
 μS/cm: microSiemen per centimeter; or micromho per centimeter (µmho/cm)
 UCMR: Unregulated Contaminant Monitoring Rule is used to collect data for contaminants that are suspected to be present in drinking water and do not have health-based standards set under the Safe Drinking Water Act

5.3

6.1

9.5

8.7

0.1

3.9

0.4

0.9

## **The Water Quality Report**

#### **Board of Directors**

- Darcy M. Burke, Division 1
- **Harvey R. Ryan,** Division 2
- > Jared K. McBride, Division 3
- > Phil Williams, Division 4
- > Andy Morris, Division 5



**Elsinore Valley Municipal Water District** 31315 Chaney Street P.O. Box 3000 Lake Elsinore, CA 92531



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